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						Course	Informat	on Shee	t & C	ourse Ou	ıtline							
							2021-2	022 Fall	Seme	ester								
Course Co			e Name							Credit			E	ECTS				
CHM111 Pre-requis			ERAL CHEMISTRY						3			5						
Language:	: English	ish			<u> </u>				Yea	ear: 2021-2022			Sem	Semester: Fall				
Weekly He	ours	Class Hours		rs	L	Laboratory Practic		um	PS			Learning Ser			ssions T		1	
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Course		Dr. Chidi Wilson NWI		NWEKV	NO/ As	sist. Prof.	Dr. Süleyman AŞl		SIR	Office Hours: Mon-11am, Tue-12pm		<b>pm</b> Thu	Thur-12pm Fri-12pm					
Lecturer/ Coordinat	H'-	E-mail address Chidi		Chidiv	wilson.nwekwo@neu.edu.tr					Online Office Hour Link: Wed - <b>12pm to 1pm</b> https://meet.google.com/ssv-qnph-ezh								
Learning Outcomes Outcomes Descriptio Course Objectives Textbooks and/or References	$s$ $rac{1}{2}$	<ul> <li>The reflection of this course, the student will be able to</li> <li>Predict physical and chemical properties of compounds based on chemical bonding, geometry and intermolecular interactions. Identify and apply recent knowledge, and analyse and solve problems in the life sciences, and understand the relationship between e life sciences, chemistry and engineering.</li> <li>Comprehend and be able to apply chemical facts, concepts, and models, and be able to excel in qualitative and quantitative problem living.</li> <li>Recognize the need for lifelong learning.</li> <li>atis course is designed as a one-semester course for freshman pharmacy students.</li> <li>udents who successfully complete this course will be able to:         <ol> <li>Understand and realize the integration of chemistry in life sciences and engineering.</li> <li>Function effectively in a medically and biologically oriented problem-solving environment.</li> <li>Develop scientific inquiry, complexity, critical thinking, mathematical and quantitative reasoning.</li> <li>Formulate meaningful conclusions according to scientific inquiry by collecting, analyzing, summarizing and interpreting laboratory data.</li> <li>Chemistry Principles and Reactions (7th edition, 2012) by William L. Masterton and Cecile N. Hurley, Brooks/Cole Cengage Learning Centre, UZEM System, and non-virtual alternative assessment tools</li> <li>basic course with emphasizing the metric system. Introduction to atomic theory, stoichiometry. The structural and physical operties of matter. Periodic relationship among elements and periodic table. Gaseous state. Thermochemistry. Energy and enthalpy. ectronic structure of atoms. Electrochemistry. Chemical bonding.</li> <li>The traditional (face-to-face) learning majorly and an interactive E-learning method</li> </ol> </li> </ul>																
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Learning Program									
Educational Tool	Amount	Student Work Load (Hours)	Educational Tool	Amount	Student Work Load(Hours)				
Course Duration	14	14*3=39							
Study Hours	14	14*3=39							
Quizz(es)	6	6*2=12							
Preparation for quizz(es)	6	6*6=36							
Final exam	1	1*2=2							
Final exam (Study hours)	1	1*12=12							
			Total		140				
		Recommende	d ECTS Credit (Total Hours / 30):	140/30 = ~ 5					